# LABORATORY-BASED PHYSICAL SCIENCE CURRICULUM

The goal of this curriculum is to increase the number of students meeting state science standards at the high school level. This can be done through this course of study in physical science, which is aligned with state science standards and meets the entrance requirements of post-secondary institutions.

Numbers in parentheses reference the Nebraska Twelfth-Grade Science Standards.

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#### **SECTION I**

#### INTRODUCTION TO SCIENCE

Concepts and Skills	Suggested Activities	Suggested
Concepts and Skins	and Resources	Assessments

**Introduction to Science** (5 days & 1 day per week throughout the rest of the course)

## 12.2.1 Scientific Inquiry

- a. Formulate questions and identify concepts that guide scientific investigations.
- b. Design and conduct scientific investigations.
- c. Use technology and mathematics to improve investigations and communications.
- d. Formulate and revise scientific explanations and models using logic and evidence.
- e. Recognize and analyze alternative explanations and models.
- f. Communicate and defend a scientific argument.

#### A. Features of Inquiry:

- 1. Engaging in Scientifically Oriented Questions
  - a. Questioning
  - b. Predicting
  - c. Forming Hypotheses
- 2. Responding to Questions using Evidence
  - a. Identifying Variables
  - b. Designing Experiments
    - i. Understand that larger well-chosen samples produce more accurate estimates of the characteristics of the total population. (12.1.2)
  - c. Making Qualitative and Quantitative Observations
    - i. Understand that measurement errors may affect results of calculations. (12.1.3)
  - d. Recording Data
- 3. Formulating Explanations from Evidence
  - a. Organizing Data
    - i. Understand that the way data are displayed affects interpretation. (12.1.2)
      - 1) Graphs
      - 2) Tables
      - 3) Calculations
    - ii. Uses of powers of ten to represent large and small numbers. (12.1.3)
      - 1) Schematics
  - b. Manipulating Data
  - c. Interpreting Evidence
    - i. Evaluate the reasonableness of answers to problems. (12.1.2)
    - ii. Understand that a correlation between two variables does not mean that either one causes the other. (12.1.2)
    - iii. Compare data for two groups by using averages and ranges of values. (12.1.3)
    - iv. Describe rate of change by comparing one measured quantity to another measured quantity. (12.1.3)
    - v. Investigate and describe how different characteristics, properties, or relationships within a system change as their dimensions increase or decrease. (12.1.3)
  - d. Creating Models
    - i. Create a physical, mental, or mathematical model to show how objects and processes are connected (12.1.2)

	Section I/Intro	duction to Science (con'
Concepts and Skills	Suggested Activities	Suggested
Concepts and Skins	and Resources	Assessments
<ul> <li>4. Connecting Explanations to Scientific Knowled a. Inferring</li> <li>b. Connecting to Existing Models <ul> <li>i. Test the usefulness of the model by conc.</li> <li>c. Defending Findings</li> <li>i. Evaluate the reasonableness of answers</li> </ul> </li> </ul>	nparing its predictions to actual obser	rvations. (12.1.2)
<ul> <li>5. Communicating and Justifying Explanations</li> <li>a. Communicating Explanations</li> <li>b. Defending Explanations</li> <li>c. Publishing</li> <li>d. Determining Applications</li> <li>e. Asking Further Questions</li> </ul>		
	<ul> <li>Logical arguments</li> <li>Science versus what is not scientific (belief based)</li> <li>Sample size and validity</li> <li>Communicate results in a scientific format</li> <li>1. A step in time     Construct a Pendulum that has a period of one second. This culminates in a class competition.</li> <li>2. Egg Bungee Jump     This application of Hooke's Law allows students to collect, graph and interpret data. Students culminate this activity by bungee jumping a raw egg as close to the floor without hitting the floor.</li> </ul>	<ul> <li>Directed Project</li> <li>Discuss examples of final projects.</li> <li>Discuss possible topics</li> <li>Formulate a project proposal</li> <li>Establish a timeline for components completion.</li> <li>Complete sample inquiry projects as a class.</li> <li>Keep a journal of project progress.</li> </ul>

- 6. 12.1.3 Change, constancy, and measurement
  - a. Uses of powers of 10 (Atoms -> the universe)
  - b. Compare data
    - i. Measurement errors and results
  - c. Rate of change by comparing one measured quantity to another measured quantity
    - i. Scientific notation
- 7. 12.1.2 Evidence, models, and explanation
  - a. Understand that the way data are displayed affects interpretation.
  - b. Evaluate the reasonableness of answers to problems.
  - c. Sample Activity: Collect data, such as mass vs. volume of different substances.
    - i. Create a physical, mental, or mathematical model to show how objects and processes are connected.
    - ii. Test the usefulness of the model by comparing its predictions to actual observations.
    - iii. Understand that a correlation between two variables does not mean that either one causes the other.
- 8. 12.6.1 Technological Design
  - a. Propose designs and choose between alternative solutions of a problem.
  - b. Implement the selected solution.
  - c. Evaluate the solution and its consequences.
  - d. Communicate the problem, process, and solution.

Section I/Introduction to Science (con't)

	Concepts and Skills	Suggested Activities and Resources	Suggested Assessments
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- 9. 12.8.2 Nature of Scientific Knowledge
  - a. Demonstrate the use of empirical standards, logical arguments, and skepticism in science.
  - b. Create scientific explanations consistent with experimental and observational evidence; make accurate predictions; strive to be logical; respect the rules of evidence; accept criticism; report methods and procedures; and make knowledge public.
  - c. Understand that all scientific knowledge is, in principle, subject to change as new evidence becomes available.

SECTION II T	ГНЕ АТОМ	<del>,</del>
Concepts and Skills	Suggested Activities and Resources	Suggested Assessments
The Atom (15 days)		L
12.3.1 The Atom  a. Investigate and describe the structure b. Investigate and explain the types of none. c. Investigate and describe the effect of	uclear reactions.	•
<ul> <li>12.8.3 Historical Contributors <ul> <li>a. Investigate and describe the contributions.</li> <li>b. Understand that changes in scientifical earlier knowledge.</li> <li>c. Understand that some advancement society.</li> </ul> </li> </ul>	ributions of diverse cultures to so	cientific knowledge an
A. Structure of the Atom:		
<ol> <li>Atomic models</li> <li>a. Thompson</li> <li>b. Rutherford</li> <li>c. Bohr</li> </ol>		
2. Electrical and nuclear forces hold atoms together	1. Electrostatic activities (Vander Graff, comb, paper, etc.)	
<ul> <li>3. Evidence supporting the existence of the electron</li> <li>a. Crooke (Cathode Rays)</li> <li>b. J.J. Thompson</li> <li>c. Properties of the electron</li> </ul>	2. Cathode Ray Demonstrations/ Video Cathode ray tubes are used to demonstrate properties of electrons, including use of a Crooke's Tube.	Lab Write Up
<ul><li>4. Evidence supporting the nucleus of the atom</li><li>a. Rutherford</li><li>b. Properties of proton</li></ul>	3. Marble Lab Model This activity models Rutherford's famous gold-foil experiments and allows students to estimate the size of the nucleus of an "atom."	Lab Write Up
<ul> <li>5. Relate subatomic particles to the atomic mass and atomic number</li> <li>a. isotopes</li> <li>b. ions</li> <li>c. decay (alpha, beta, gamma)</li> <li>d. Contributions of Marie Curie and Becquerel</li> <li>e. Neutron</li> </ul>	<ul> <li>4. Half Life Simulation Pennies and pizza boxes are used to collect data. In a closed pizza box the coins are tossed and all "heads" are removed and the process is repeated through 4-5 half-lives.</li> <li>5. Transmutation Game <i>Interactions Course</i> 3, Glencoe, Activities Book Shows half-life and element it</li> </ul>	

decay.

decays to. Playing cards and

game boards are made by students are used to practice learning the new isotopes formed by alpha and beta

Section II/The Atom (con't)

		ii ii, i ne i teom (con t)
Concepts and Skills	Suggested Activities	Suggested
Concepts and Skins	and Resources	Assessments
6. 12.5.3 Predict when rocks were	6. Relative/Absolute Dating of Rocks	Lab Write Up
formed by using known decay	(www.ucmp.berkeley,edu/fosrec/McKinney.html)	
rates of radioactive isotopes in	This activity uses M&M's to determine half-	
rocks.	lives. It has block diagrams to show relative	
	ages of events and to introduce absolute	
	dating. These are student centered activities.	
	(Rated for grades 8-12)	

SECTION III	MATTER	
Concepts and Skills	Suggested Activities and Resources	Suggested Assessments
Matter (30 days)		
<ul> <li>12.3.1 The Atom</li> <li>a. Investigate and describe the structure</li> <li>b. Investigate and explain the types of n</li> <li>c. Investigate and describe the effect of</li> </ul>	uclear reactions.	•
<ul> <li>A. 12.3.2 Structure and Properties of Matter</li> <li>• Investigate and understand that atoms inter</li> <li>• Investigate and explain the periodic table chemical properties.</li> <li>• Investigate and describe how the structure</li> <li>• Investigate and describe how the interactio and chemical properties.</li> </ul>	act with one another by transferring of of elements in terms of repeating of an atom determines the chemical p	or sharing electron. patterns of physical an
1. Investigate and use changes in energy to	Predict the Properties of Element	Lab Write Up
explain the differences among states of matter.  a. Phase changes  2. Electron configuration  3. Transferring and sharing of electrons  a. Bonding  i. Ionic  ii. Covalent  iii. Metallic  4. Periodic Table  a. Investigate properties of some elements  b. Examine patterns in the periodic table  c. Valence electrons	Students use periodic data of known elements within the group to predict the molecular weight, density, and mass of element 114. Students then search the internet to compare their predictions to the most current information.  Lab 7 Relationships Among Elements Laboratory Manual Science Interactions Course III, Glencoe  Please, Set a Table! Science Discovery Activity	Internet Research
d. Properties of metals versus nonmetals	Science Interactions Course III, Glencoe  Ready to Use Chemistry Activities Handwerker Center for Applied Research in Education ISBN: 0-87628-438-1  Shows a periodic table, click on element to find history and properties http://education.jlab.org/itselemental/	

Interactive table of elements http://www.webelements.com/webelements/scholar/index.html

Section III/Matter (con't)
Suggested

B. 12.3.3 Understanding of Chemical Reactions  • Investigate and describe common chemical reactions.  • Investigate and describe the change of energy as a result of chemical reactions.  • Investigate and give examples of how systems tend to become more disorderly over time. (12.3.5)  • Investigate and describe how electrons are involved in bond formation during chemical reactions.  • Investigate and describe the bonding of carbon atoms in chains and rings to produce compounds essential to life. (12.3.2)  • Investigate and describe the factors influencing the rates of chemical reactions including catalysts.  1. Common chemical reactions  a. Synthesis  b. Decomposition  c. Single replacement  d. Double replacement  e. Combustion  3. Endothermic versus exothermic reactions  a. Entropy  Activity 14: Sour Chemistry  Real World Math with the CBL  System.  Brueningsen, etal, 1994, ISBM 1-  886309-03-5  Lab Summary  Balancing Equations  Work Sheets  Activity 14: Sour Chemistry  Real World Math with the CBL  System.  Brueningsen, etal, 1994, ISBM 1-  886309-03-5  Lemon juice is neutralized with effervescent antacids. Data is collected using a pH probe and CBL system. Students analyze		Suggested Activities	Suggested
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<ul> <li>Investigate and describe the factors influencing the rates of chemical reactions including catalysts.</li> <li>Common chemical reactions         <ul> <li>a. Synthesis</li> <li>b. Decomposition</li> <li>c. Single replacement</li> <li>d. Double replacement</li> <li>e. Combustion</li> </ul> </li> <li>Balancing equations</li> <li>Balancing Equations         <ul> <li>Work Sheets</li> </ul> </li> <li>Activity 14: Sour Chemistry         <ul> <li>Read World Math with the CBL</li> <li>System,</li> <li>Brueningsen, etal, 1994, ISBM 1-886309-03-5</li> <li>Lemon juice is neutralized with effervescent antacids. Data is collected using a pH probe and CBL system. Students analyze</li> </ul> </li> </ul>	<ul> <li>Investigate and describe the bonding of car</li> </ul>	bon atoms in chains and rings to produce	duce compounds essential
1. Common chemical reactions a. Synthesis b. Decomposition c. Single replacement d. Double replacement e. Combustion  2. Balancing equations 3. Endothermic versus exothermic reactions a. Entropy 4. Catalysts, Temperature, Physical Surface Area Exposure, Concentration  Students will move from station to station to station to carry out each of the common types of chemical reactions. Students will write formulas for the reactions they observe.  Activity 14: Sour Chemistry Real World Math with the CBL System, Brueningsen, etal, 1994, ISBM 1- 886309-03-5 Lemon juice is neutralized with effervescent antacids. Data is collected using a pH probe and CBL system. Students analyze	to life. (12.3.2)		
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b. Decomposition c. Single replacement d. Double replacement e. Combustion  2. Balancing equations  3. Endothermic versus exothermic reactions a. Entropy  4. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  4. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  5. Decomposition c. Single replacement common types of chemical reactions. Students will write formulas for the reactions they observe.  6. Activity 14: Sour Chemistry 6. Real World Math with the CBL 6. System, 6. Brueningsen, etal, 1994, ISBM 1- 6. Balancing Equations 6. Work Sheets  8. Work Sheets  8. Lemon juice is neutralized with effervescent antacids. Data is collected using a pH probe and CBL system. Students analyze	Common chemical reactions	Reaction Types	Lab Summary
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c. Single replacement d. Double replacement e. Combustion  2. Balancing equations  3. Endothermic versus exothermic reactions a. Entropy  4. Catalysts, Temperature, Physical Surface Area Exposure, Concentration  Common types of chemical reactions. Students will write formulas for the reactions they observe.  Activity 14: Sour Chemistry Real World Math with the CBL System, Brueningsen, etal, 1994, ISBM 1- 886309-03-5 Lemon juice is neutralized with effervescent antacids. Data is collected using a pH probe and CBL system. Students analyze	b. Decomposition	to station to carry out each of the	Balancing Equations
d. Double replacement e. Combustion  2. Balancing equations  3. Endothermic versus exothermic reactions a. Entropy  4. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  4. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  4. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  4. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  5. Students will write formulas for the reactions they observe.  4. Activity 14: Sour Chemistry  8. Real World Math with the CBL  8. System,  8. Brueningsen, etal, 1994, ISBM 1-  8. Segondary 1-  8. Students will write formulas for the reactions they observe.  4. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  6. Combustion  6. Combustion  7. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  8. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  8. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  8. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  8. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  8. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  8. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  8. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  9. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  9. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  9. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  9. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  9. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  9. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  9. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  9. Catalysts, Tempe			
e. Combustion  2. Balancing equations  3. Endothermic versus exothermic reactions a. Entropy  4. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  Surface Area Exposure, Concentration  formulas for the reactions they observe.  Activity 14: Sour Chemistry Real World Math with the CBL System, Brueningsen, etal, 1994, ISBM 1-886309-03-5 Lemon juice is neutralized with effervescent antacids. Data is collected using a pH probe and CBL system. Students analyze			
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3. Endothermic versus exothermic reactions a. Entropy  4. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  System, Brueningsen, etal, 1994, ISBM 1-886309-03-5  Lemon juice is neutralized with effervescent antacids. Data is collected using a pH probe and CBL system. Students analyze		Activity 14: Sour Chemistry	ļ ļ
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4. Catalysts, Temperature, Physical State, Surface Area Exposure, Concentration  886309-03-5  Lemon juice is neutralized with effervescent antacids. Data is collected using a pH probe and CBL system. Students analyze			
Surface Area Exposure, Concentration  Lemon juice is neutralized with effervescent antacids. Data is collected using a pH probe and CBL system. Students analyze	4. Catalysts. Temperature. Physical State.		
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collected using a pH probe and CBL system. Students analyze	1 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1,		
CBL system. Students analyze			
		data using a math model.	

SECTION IV ENERGY		
Concepts and Skills	Suggested Activities and Resources	Suggested Assessments
Energy (25 days)		
<ul> <li>A. Types of Energy</li> <li>Kinetic (energy due to motion)</li> <li>Potential (energy due to position)</li> <li>Mechanical (kinetic + potential)</li> <li>Thermal</li> <li>Chemical</li> <li>Electromagnetic</li> </ul>		
• Nuclear		
	yes possess and transfer energy.  gth and frequency of waves are inversely  nergy of waves can be changed into other	er forms of energy can be
• Investigate and understand that the energy of waves can be changed into other forms of energy can transformed into energy, just as other forms of energy can be transformed into wave energy.  1. Wave characteristics  a. Wavelength  b. Frequency  c. Amplitude (crest and trough)  2. Wave types  a. Transverse  b. Longitudinal  b. Longitudinal  3. Wave Activity  Activity Two: Sounds in Strings  I Lab Write up  Students use a string with varying lengths and masses to create numerous sound frequencies.  Students can create a musical presentation based on their findings.  (http://www.Its-About-Time.com)  Absorb Physics for GCSE  This site has some interactive demonstrations on waves. It requires students to use graphical data that students make calculations from to input solutions into the computer. This is a .com site that sells a more		
<ul> <li>C. 12.3.6 Interactions of Energy and Matt</li> <li>• Investigate and understand that atoms discussion).</li> <li>1. Spectrum analysis; electromagnetic spectrum</li> </ul>	s or molecules can be identified by spec	

- a. Sloan Digital Sky Survey Emission
  b. Spectra for stars are shown. Students can use the data available to classify specific types of stars based on spectral data
  c. <a href="http://cas.sdss.org">http://cas.sdss.org</a>

Section IV/Energy (con't)

		I	Section IV/Energy <i>(con't)</i>
	Concepts and Skills	Suggested Activities	Suggested
	•	and Resources	Assessments
D.	12.3.5 Conservation of Energy and Increas		
	• Understand that the total energy in the univ		troyed.
	• Investigate and distinguish between kinetic		40
	• Investigate and describe heat transfer in ter	rms of conduction, convection, and ra	diation.
	Law of Conservation of Energy		
	a. Mechanical		
	o. Thermal (conduction, convection, radiation)		
Ε.	12.5.1 Understanding energy in the earth s		a dagar and arritational
	• Investigate and distinguish between intern		
	energy) and external sources of energy (e earth system.	.g., the sun), and explanning now bo	our provide energy to the
	<ul> <li>Investigate and explain how the outward to</li> </ul>	ransfer of earth's internal heat drives	convection in the mantle
	that propels the plates comprising the earth		convection in the mantic
	<ul> <li>Investigate and explain how global clim</li> </ul>		fer from the sun and is
	influenced by dynamic processes (e.g., cl		
	(e.g., the position of mountain ranges and o		on) and state conditions
1. F	Energy in the earth system		
	a. Core-mantle-crust structure (review)		
	o. Thermal energy in Earth's interior drives		
	the geosphere		
	i. Evidence for interior heat		
	1) Volcanoes		
	2) Deep boreholes: deeper = hotter		
	ii. Origin of heat in the interior		
	1) Residual from bombardment		
	(mechanical energy, friction)		
	2) Radioactive decay (25% of heat)		
	iii. Consequences of hot interior		
	1) Heat transfer through mantle to		
	a) Plate tectonics: convection		
	mantle moves the plates		
	carrying oceans and continents		
	b) Geological activity (volcanoes,		
	earthquakes) at plate boundaries		
	e. Solar energy (external) drives the		
	atmosphere		
	i. Radiant heat (infrared)		
	1) Absorption by land and water		
	2) Differential heating of surface		
	leads to convection of the atmosphere		
	a) Earth's wind (climate) belts		
	b) Coriolis effect		
	c) Influence of water bodies and		
	land masses (rain shadow,		
	differential absorption, wind		
	deflection, etc.)		
	d) Convection and cloud formation		
<u> </u>	e) Cloud classification		

#### SECTION V

## MOTION AND FORCES

SECTION V MOTI	ON AND FORCES	
Concepts and Skills	Suggested Activities	Suggested
Concepts and Skins	and Resources	Assessments
Motion and Forces (20 days)		
12.3.4 Motions and Forces	4	
a. Investigate and understand of force o	2	
b. Investigate and understand gravity as		
<ul><li>c. Investigate and understand electrical</li><li>d. Investigate and describe an electric fi</li></ul>		
A. 12.8.3 Historical Contributors	erd a magnetic field, and the interaction	on between them.
<ul> <li>Investigate and describe the contributions</li> </ul>	of diverse cultures to scientific know	vledge and technological
investigate and desertor the contributions	of diverse cultures to scientific know	vicage and technological
<ul> <li>Understand that changes in scientific knowledges</li> </ul>	owledge evolve over time and almos	t always build on earlier
knowledge.	with the control of the control with the control of	v with wy 5 contra off contract
<ul> <li>Understand that some advancements in ser</li> </ul>	ience and technology have long-lastin	g effects on society.
1. Aristotle – Fire, Air, Earth, Water		
2. Copernicus – Heliocentric Theories		
3. Keppler – Planetary Motion		
4. Galileo – Experimental Verification and		
Testing of Hypotheses		
5. Newton – Mechanics		
6. Einstein – Special and Relative Motion;		
Photoelectric Effect		
B. Newton's Law		
1. Law of inertia (1 <sup>st</sup> )	TOPS Module 21 Motion	
2 T C 4 4 1 4 (2nd)	Marson, 1979	1
2. Law of constant acceleration (2 <sup>nd</sup> )	This module contains seve	
2 Action reaction (2 <sup>rd</sup> )	activities testing uniform motion	
3. Action – reaction (3 <sup>rd</sup> )	tracking collisions, and Newton second law.	1 8
	Active Physics Sports	
	Eisenkraft, 1998, ISBN 1-891629-04-2	2
	This text has a myriad of activities	
	students based on motion and sporti	
	activities.	
	Physics with Computers	
	Gastineau, et.al., Vernier Softwa	re,
	1998	
	ISBN 0-918731-99-2	
	This source uses interfacing syste	
	to provide data collection and analy	
	activities to test numerous forms motion.	01
	Middle School Science with Computers	,
	Volz and Sapatka, Vernier Softwa	
	2000	
	ISBN 1-929075-08-1	
	This reference integrates interfaci	ng
	systems and lab activities to test	
	variety of physics activities from t	
	greenhouse effect to ocean flo	oor
	mapping to various forms of motion	l.

Section V/Motion and Forces (con't)

Concepts and Skills	Suggested Activities and Resources	Suggested Assessments
C. Gravity		
1. Force of attraction between any two objects		
2. Distinction between weight and mass		
D. Electromagnetic Force		
1. Like and unlike charges (+ and -)	Review electrostatic activities	
2. May be attractive or repulsive		
3. Holds atoms together (electrons— and nucleus+)		
4. 12.3.6—Understand that electromagnetic waves occur when a charged object accelerates.		
5. Interactions between electrical and magnetic fields		

# ORIGIN OF THE UNIVERSE

	Concepts and Skills		Suggested Activities	Suggested
•			and Resources	Assessments
12.	5.4 Origin of the Universe (8 days)		4	
	a. Describe and analyze the Big Bang th			1
	b. Describe various types of galaxies in	clud	ing spherical, elliptical, and irregi	ular
	c. Describe the life cycle of a star.			- ·
Α.	Big Bang	1.	Sloan Digital sky Survey	Research paper
	1. Evidence for expanding universe		(cas <u>.sdss.org</u> )	- 1 -
	a. red shift		By starting at the basic level	Lab Report
	b. Hubble's Law		several student/ teacher	
			activities exist to challenge	Class presentations
В.	Galaxy Formation		students, such as a scavenger	
	1. Classification of galaxy types		hunt through the origin of	
			galaxies and the universe.	
C.	Birth and Death of Stars			
	1. Hertzsprung-Russell diagram	2.	Project Earth Science Astronomy	
	2. Spectral analysis		Smith, NSTA Press, 2001,	
			<u>ISBN 0-87355-108-7</u> "Activity	
			6"	
			Using vermiculite and water	
			this model demonstrates the	
			accretion theory when rotating	
			interstellar clouds of gas and	
			dust collapsed under its own	
			gravity.	
		3.	Celery seeds and a pan of water	

SECTION VII CYCLES

Concepts and Skills	Suggested Activities	Suggested				
Concepts and Skins	and Resources	Assessments				
Cycles (16 days)						
A. 12.5.2 Understanding of Geochemical Cycles (5 days)						
· Investigate and diagram how elements and compounds on earth move among reservoirs in the solid						
earth, oceans, atmosphere, and organisms as part of geochemical cycles. (Water, Rock, Carbon						
Nitrogen)						
Reinforce cycle concept	Rock Around the Clock"					
a. Investigate and understand that atoms and						
molecules cycle among living and	Project Earth Science Geology					
nonliving components of the biosphere.						
	Ford, 2001, "Activity 13:					
2. Rock Cycle	Students investigate the rock					
	cycle using crayons. Shaved					
	crayons are used to model					
	chemical/mechanical weathering,					
	as well as sedimentary rock					
	formation					

#### B. 12.1.1 Develop an understanding of systems, order, and organization

- Predict and evaluate how change within a system affects that system
- Design solutions to problems identified within a system

# C. 12.7.6 Role of science and technology in local, national, and global challenges. (10 days)

- Understand that knowledge of basic concepts about scientific and technological challenges should precede active debate.
- Investigate and understand that social issues and challenges may affect advancements in science and technology.
- Understand that science and technology are essential social enterprises that indicate what could happen, but not what should happen.
- Social perspectives of global warming issue
- Role of greenhouse gases (CO<sub>2</sub>, CH<sub>4</sub>, H<sub>2</sub>O)
  - 1. Evidence for increase and industrial influence
  - 2. Evidence for global temperature increase
  - 3. Examine and understand the societal, cultural, and personal beliefs that influence scientists.
  - 4. Alternatives to fossil fuels: nuclear, renewable (solar, wind, hydro, biomass)
- Carbon sources (combustion of fossil fuels, volcanoes, respiration
  - 1. Short term (human time scale)
- Carbon sinks (burial of biomass coal, limestone, etc.)
  - 1. Long-term (geological time scale)

# SECTION VIII EARTH'S HISTORY

Concepts and Skills	Suggested Activities and Resources	Suggested Assessments			
Earth's History (17 days)					
12.5.3 Origin of the earth system					
a. Contrast the early earth with the plan					
<ul> <li>Investigate and estimate geologic time by observing rock sequences and using fossils to correlate the sequences at various locations.</li> </ul>					
organisms affect the ongoing evolution					
1. Explanations for plate movements (Pangaea	Video showing continent motions				
and older continent-ocean configurations)  a. Convection in the mantle					
	www.accessexcellence.com (General Science)				
<ul><li>b. Origin of the magnetic field</li><li>c. Evidence for plate movement</li></ul>	(General Science)				
d. Plate tectonics theory	Project Earth Science				
d. Trace tectories theory	Troject Earth Science				
	Ford, 2001 "Activity 6: Convection"				
	(A convection cell in water is				
	constructed by students. This				
	model demonstrates how				
	convection plays a role in plate				
	tectonics.)				
2. Earth history and the rock record	Earth & Astronomical Science				
a. Geologic time scale	Activities for Grades 5-12				
b. Relative dating	ISBN: 0-87628-445-4				
i. Superposition					
ii. Unconformities	The Center for Applied Research				
iii. Fossils	in Education				
c. Numerical dating					
i. Radiometric					
ii. Tree rings					
d. Interactions among earth systems in					
geologic time					